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APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. 09/699,451 10/31/2000 Kersti Strandqvist P15484US00 8608 466 EXAMINER 7590 07/26/2004 YOUNG & THOMPSON ANDERSON, CATHARINE L 745 SOUTH 23RD STREET 2ND FLOOR ART UNIT PAPER NUMBER ARLINGTON, VA 22202 3761

DATE MAILED: 07/26/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

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Applicant(s)				
STRANDQVIST, KERSTI				
Art Unit				
3761				
orrespondence address				
S) FROM				
ely filed				
will be considered timely. the mailing date of this communication. O (35 U.S.C. § 133). may reduce any				
secution as to the merits is 3 O.G. 213.				
Examiner. 37 CFR 1.85(a).				
ected to. See 37 CFR 1.121(d).				
Action or form PTO-152.				
-(d) or (f)				

	Application No.	Applicant(s)	Applicant(s)	
Office Action Summary	09/699,451	STRANDQVIST,	STRANDQVIST, KERSTI	
	Examiner	Art Unit		
	C. Lynne Anderson	3761		
The MAILING DATE of this commun. Period for Reply	ication appears on the cover sheet w	vith the correspondence a	ddress	
A SHORTENED STATUTORY PERIOD FOR THE MAILING DATE OF THIS COMMUNI - Extensions of time may be available under the provisions after SIX (6) MONTHS from the mailing date of this community of the period for reply specified above is less than thirty (3). If NO period for reply is specified above, the maximum states a period for reply within the set or extended period for reply Any reply received by the Office later than three months a earned patent term adjustment. See 37 CFR 1.704(b).	CATION. of 37 CFR 1.136(a). In no event, however, may a nunication. O) days, a reply within the statutory minimum of thi atutory period will apply and will expire SIX (6) MO will, by statute, cause the application to become A	reply be timely filed rty (30) days will be considered time NTHS from the mailing date of this BANDONED (35 U.S.C. § 133).		
Status				
1) Responsive to communication(s) file	d on <u>07 A<i>pril</i> 2004</u> .			
2a)⊠ This action is FINAL.	FINAL. 2b) This action is non-final.			
3) Since this application is in condition closed in accordance with the practice	·	• •	e merits is	
Disposition of Claims	·			
4) ⊠ Claim(s) <u>13-24</u> is/are pending in the 4a) Of the above claim(s) is/are 5) □ Claim(s) is/are allowed. 6) ⊠ Claim(s) <u>13-24</u> is/are rejected. 7) □ Claim(s) is/are objected to. 8) □ Claim(s) are subject to restrict	re withdrawn from consideration.			
Application Papers				
9) ☐ The specification is objected to by the	e Examiner.			
10) The drawing(s) filed on is/are:	a) accepted or b) dojected to	by the Examiner.	•	
Applicant may not request that any object	ction to the drawing(s) be held in abeya	nce. See 37 CFR 1.85(a).		
Replacement drawing sheet(s) including 11) The oath or declaration is objected to	•	• • •	• •	
Priority under 35 U.S.C. § 119				
2. Certified copies of the priority3. Copies of the certified copies of application from the Internation	documents have been received. documents have been received in a of the priority documents have been nal Bureau (PCT Rule 17.2(a)).	Application No n received in this Nationa	ıl Stage	
* See the attached detailed Office action	n for a list of the certified copies no	t received.		
Attachment(s)				
1) Notice of References Cited (PTO-892)		Summary (PTO-413)		
 Notice of Draftsperson's Patent Drawing Review (PTO-948) □ Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) □ Paper No(s)/Mail Date 5) □ Notice of Informal Patent Application (PTO-152) Paper No(s)/Mail Date 			^T O-152)	

U.S. Patent and Trademark Office PTOL-326 (Rev. 1-04)

Art Unit: 3761

DETAILED ACTION

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 13-17, 19, and 21-24 are rejected under 35 U.S.C. 103(a) as being unpatentable over Assarsson et al. (3,901,236) in view of Smith (5,328,939).

With respect to claims 13 and 14, Assarsson discloses all aspects of the claimed invention but remains silent as to the method of crosslinking the superabsorbent material. Assarsson discloses an absorbent structure in an absorbent article, the absorbent article being a diaper, incontinence guard, and a sanitary napkin, as described in column 1, lines 7-9, and column 7, lines 64-66. The structure comprises a porous material comprising fibers and a superabsorbent material, as disclosed in column 1, lines 9-11. The structure contains at least 50% superabsorbent material, as disclosed in column 5, lines 22-24. The superabsorbent material is in the form of particles, a shown in figure 7, and may be crosslinked by covalent or ionic bonds, as disclosed in column 3, lines 37-40. Suitable polymers for the superabsorbent material are disclosed in column 3, line 41 to column 4, line 12, and includes poly(vinylpyridine).

Smith discloses a method of forming a superabsorbent material for use as an absorbent, as disclosed in column 2, lines 54. The superabsorbent material is

Application/Control Number: 09/699,451

Art Unit: 3761

produced by initially ionically crosslinking a polymer, such as poly(vinylpyridine), as disclosed in column 3, lines 6-10). The polymer is crosslinked with an ionic crosslinking agent comprising a polyvalent metal ion, as disclosed in column 4, lines 19. The superabsorbent material is then covalently crosslinked, as disclosed in column 4, lines 59-63. The method disclosed by Smith results in a superabsorbent material that provide improved strength and are highly stable, as described in column 2, lines 49-51.

It would therefore be obvious to one of ordinary skill in the art at the time of invention to produce the superabsorbent material of Assarsson by the method taught by Smith to provide the absorbent structure with improved strength and stability.

With respect to claim 15, the superabsorbent is poly(vinylpyridine), which is a polymer with an anionic functional group.

With respect to claim 16, the superabsorbent material disclosed by

Assarsson in column 3, line 46, is polyacrylic acid, which has functional carboxy

groups and the same structure as polyacrylate.

With respect to claim 17, the covalent crosslinking agents disclosed by Smith in column 4, line 63 to column 5, line 36, have cations which will bond to anionic functional groups.

With respect to claim 19, the polyvalent metal ion is aluminum or zinc, as disclosed by Smith in column 4, line 19.

With respect to claim 21, the superabsorbent material comprises crosslinking along its surface, and is therefore surface crosslinked.

Application/Control Number: 09/699,451

Art Unit: 3761

With respect to claim 22, the superabsorbent material is homogenously mixed with the ionic crosslinking agent, and therefore is substantially homogeneously crosslined, as disclosed by Smith in column 7, lines 36-43.

With respect to claim 23, the structure contains 75% by weight of superabsorbent material, as disclosed by Assarsson in column 5, lines 22-23.

With respect to claim 24, diapers, sanitary napkins, and incontinence devices are well-known to comprise a liquid pervious topsheet, a liquid pervious backsheet, and an absorbent structure located therebetween.

Claims 18 and 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Assarsson et al. (3,901,236) in view of Smith (5,328,939) as applied to claim 15 above, and further in view of Hutchins et al. (5,957,203).

Assarsson, as modified by Smith, discloses all aspects of the claimed invention with the exception of the cationic crosslinking agent being aluminate ion.

Hutchins discloses in column 3, lines 39-45, the use of aluminate ion, which comprises the polyvalent metal ion aluminum, as a cationic crosslinking agent. The crosslinking agent disclosed by Hutchins provides the crosslinked polymer with improved gel qualities, as described in column 1, line 33 to column 2, line 15.

It would therefore be obvious to one of ordinary skill in the art at the time of invention to use aluminate ion as the cationic crosslinking agent, as taught by Hutchins, in the method of forming an absorbent structure disclosed by Smith.

Response to Arguments

Applicant's arguments filed 7 April 2004 have been fully considered but they are not persuasive.

In response to applicant's argument that the references fail to show certain features of applicant's invention, it is noted that the features upon which applicant relies (i.e., a high distribution capacity at repeated wettings) are not recited in the rejected claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

In response to applicant's argument that the teachings of Smith are not applicable to the invention of Assarsson because Smith discloses a rigid, gasabsorbing material and makes no mention of "superabsorbent materials," it is noted that Smith discloses in column 2, line 67 through column 3, line 11, a "hydrogel polymer" that absorbs water. Hydrogel polymers are well known in the art to be in the category of superabsorbent materials. Further, Smith discloses the use of the same polymers as Assarsson, but teaches that ionically crosslinking the polymers prior to covalently crosslinking the polymers results in improved stability of the absorbent material. Therefore, the modification of Assarsson in view of Smith is proper.

In response to applicant's argument that the teachings of Hutchins are not applicable to the invention of Assarsson as modified by Smith, it is noted that Smith discloses in column 6, lines 24-27, subjecting the crosslinked polymer to

Application/Control Number: 09/699,451

Art Unit: 3761

high heats. Therefore, the crosslinking agent must be able to withstand high heats, and the teaching of Hutchins is relevant.

Conclusion

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to C. Lynne Anderson whose telephone number is (703) 306-5716. The examiner can normally be reached on Monday through Friday.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, John Calvert can be reached on (703) 305-1025. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Art Unit: 3761

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Cla

July 19, 2004

JOHN SCALVERT
SUPERVISORY PATENT EXAMINER
TECHNOLOGY CENTER 3700